



2<sup>nd</sup>Annual Illinois Basin Coal Symposium

## Combustion Optimization by Application of the Fundamentals as it Relates to Illinois Basin Coal

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# Goals of this Talk:

- 1. Discuss How and Why Furnace Combustion Optimization is important.
- 2. Present the Storm "13 Essentials of Combustion Optimization in PC Units.
- 3. Provide a review of the consequences of "Not Optimizing Furnace Inputs"
- 4. Describe How Storm Recommends Optimizing the Furnace Inputs. (Both Air & Fuel)
- 5. Review Large Utility Boiler Furnace Combustion.
- 6. Briefly Cover Why Some boilers are more "Forgiving" than others.



#### FYI, Storm Technologies, Inc. Business is **Improving Overall Coal Plant Performance**



# **Typical Opportunities for Improvement** Capacity, Reliability, Heat Rate, Environmental and Fuels Flexibility







### **The Combustion of Solid Carbon**

• The combustion of solid carbon or "char" must be completed in a very short time.

• If air and carbon are not well mixed, combustion requires even more time. From the time a coal particle enters the furnace, it spends approximately 1 to 1.5 seconds above it's ignition temperature of 1,400 degrees F.

• Most of this time is used for burning of the carbon.



Carbon Char Burn-Out Requires 3 Essential Elements



This graph illustrates typical time requirements for combustion of coal. These times will vary with different coals & firing conditions but the combustion of carbon always requires the most time.



## Low NO<sub>X</sub> Firing Evolution Challenges



## **Combustion Airflow Distribution & Control**





### **Total Airflow Measurement Example**



### The Furnace Exit Gas Temperature (FEGT)





High furnace exit gas temperatures can contribute to overheated metals, such as these superheater alignment castings that only lasted 1 year due to greater than 2,500°F. furnace exit gas temperatures.

### **Furnace Exit HVT Testing**



### **Tube Metal Thermocouples**





Installation of PSH Tube Metal Thermocouples

The flue gas Bulk temperatures typically coincide with "Hot" tube circuits



Primary Super-Heat (PSH) Element Tube Metal Thermocouple Installation Progress

## **Online FEGT Monitors**



## Slagging or Fouling, in our experience, is likely to be from one of Five Common Root Causes:

- Airflow Imbalances
- Oxygen Deficiency in the Furnace
- Fuel Flow Imbalances
- High Primary Airflows
- Poor Fuel Fineness



### **Precise Combustion Air Staging**



## **Boiler Testing & Tuning First: Identify Opportunities**





### Excessive de-superheating spray flows & heat rate



### **Furnace Exit Gas Profiles**



Minimization of Reducing atmospheres at the furnace exit is the Key to optimizing flue gas temperatures and reducing slag bridging, heavy levels of secondary combustion and hot tube circuits.



## **Secondary Combustion (video)**



### <u>Typical Flue Gas Stratifications & Flue Gas</u> <u>Temperatures - Velocities</u>



## Review of the Solution for the Associated Challenges with High Sulfur, Illiniois Basin Coal

# Three Basic Concerns to Mitigate:

- Waterwall Wastage
- Slagging from Secondary Combustion
- SCR or Airheater Fouling



<u>High Sulfur</u> <u>vs.</u> Low Sulfur Coal





	High	Low	
Ultimate Analysis	Slagging	Slagging	
	Coal	Coal	
Malatura	C 00	2.2	0/ 64
Wolsture	6.09	2.2	% Dy WL
Carbon	63.25	12.1	% Dy WL
Hydrogen	4.32	4.7	% DY WL
Nitrogen	1.37	1.27	% Dy Wt.
Sulfur	3.81	0.76	% Dy Wt.
Ash	15.63	13.4	% by wt.
Oxygen	5.53	4.97	% by wt.
Ash Fusion Temperatures			
Reducing			
Initial Deformation	1,955	2750+	°F
Softening	2,180	2750+	°F
Hemispherical	2,290	2750+	°F
Fluid	2,400	2750+	°F
	_,		
Oxidizing			
Initial Deformation	2,440	2750+	°F
Softening	2,515	2750+	°F
Hemispherical	2.585	2750+	°F
Fluid	2,660	2750+	°F
Manual Ash, Assherin			
Silicon Diovide	45.0	50.6	% by wt
Aluminum Oxide	40.9	27 42	% by wt
Titopium Oxide	20.0	1 24	70 Dy WL. 9/ by wt
Iranium Oxide	0.90	1.34	% by WL
Colorium Ouide	20.94	4.07	76 Dy WL
Calcium Oxide	1.30	0.62	% Dy Wt.
wagnesium Oxide	0.73	0.75	% Dy Wt.
Potassium Oxide	2.13	2.47	% Dy Wt.
Sodium Oxide	0.21	0.42	% by wt.
Sulfur Trioxide	0.91	0.99	% by wt.
Phosphorous Pentoxide	0.3	0.42	% by wt.

### **The Water Wall Corrosion Process**

- > Incomplete combustion causes sodium and potassium in the coal to become oxides
- Sulfur from the coal combines with oxygen to form sulfur dioxides.
- These compounds are then deposited on the tube surfaces.
- The deposited sulfur compounds combine with the sodium or potassium oxides to form pyrosulfates.
- > When the pyrosulfates, carbon, and iron combine, wastage will occur.



## H<sub>2</sub>S & NO<sub>X</sub> Correlation

The difficulties with lowering the production of  $H_2S$  is that it is closely tied to the CO level in the furnace and inversely related to  $NO_X$  production. Meeting  $NO_X$  limits while simultaneously lower harmful levels of  $H_2S$  require well staged and measured airflow. As you can see below –





## Relationship of Poor Fineness w/ Water Wall Wastage





Microscopic Investigation of Deposits Source: Rod Hatt, CCI

Poor Fineness will not only result in poor distribution, but also heavier Iron Concentration in the Ash; **High Iron + Reducing Atmosphere = Trouble** 



### **Balancing the fuel lines by Clean air**

Balance the fuel line system resistances by clean air testing. Using the STORM Two Team, Dual Traverse Method, to achieve resistance within 2% for all pipes.



## <u>Key Parameters for Characterizing</u> <u>Mill Performance & Capacity</u>

- Coal HGI & Moisture
- Coal Fineness
- Primary air flow Accuracy
- Air & Fuel Control Across the load range
- Input Power requirements
- Mill Outlet Temperature
- Pyrite/Coal Rejects



### **Dirty Airflow Testing & Isokinetic Coal Sampling**

- Ascertain relative pipe to pipe fuel balance.
- Quantify individual fuel line air to fuel ratios
- Quantify pulverizer air to fuel ratio
- Þ Quantify individual fuel line velocity and airflow
- Ascertain pipe to pipe airflow balance
- Quantify fuel line temperature and static pressure
- Obtain representative fuel samples for coal fineness analysis







### Coal Fineness Analyses

Fuel line fineness shall be 75% or more passing a 200 mesh screen. 50 mesh particles shall be less than 0.1%



## **<u>Average Collected Particle Size</u>** (from Isokinetic Coal Sampling)

60% thru 200 mesh vs. 80% thru 200 mesh, yields a 85.7% difference in the particle surface area (mm<sup>2</sup>)



## Pulverizer Optimization is Not Optional First Affect of Fuel Fineness on NO<sub>X</sub>

"Release of Fuel Bound Nitrogen in the De-Volatilization Zone"



# **<u>Performance Testing Data</u>** (Before & After Performance Improvements via Isokinetic Coal Sampling)

Note: Coal is 1,000 times more dense than air. The finer the product the better the distribution (as finer coal acts more like a fluid or gas).



### Effects of poor coal fineness vs. Good coal fineness **Mechanical Synchronization With Velocity Vectors**



### Primary air/fuel ratio shall be accurately measured & controlled when above minimum



Optimum Primary Airflow Contributes to Best Heat Rate Operation



High Tempering Airflow Bypasses the Air Heater and contributes to a less desirable "X" Ratio. Therefore, the mills must be optimized to insure that optimum performance is compatible with a desirable air-fuel ramp

### STORM<sup>®</sup> Flyash Samplers (Traditional)





### **Flyash Analysis**





# Getting **RESULTS!**

 Test to Identify and Quantify Opportunities
Combine Test Data with Outage Planning (Performance Driven Maintenance)
Implement Improvements
Tune, Balance and Calibrate Airflows and Fuel Flows
Practice Performance Preservation Throughout the Year



### **Performance Driven Maintenance Techniques**



## **Thank You!**

Any Questions? Wishing you a very good year of high capacity factor, low generation cost, high reliability and environmentally sustainable power generation.

